## A Novel Synthesis of the Macrocyclic Spermidine Alkaloid  $(+)$ - $(S)$ -Dihydroperiphylline

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A novel, short, and highly stereoselective synthesis of the macrocyclic spermidine alkaloid  $(+)$ - $(S)$ dihydroperiphylline (15) is described. The key synthetic steps were the stereoselective addition of the chiral amine 1 to the cinnamate 2 and cyclization of the bis[toluene-4-sulfonamide] precursor 12 in the presence of  $Cs<sub>2</sub>CO<sub>3</sub>$  as a template. Unambiguous assignments of the signals in both the  $^1H$ - and  $^{13}C$ -NMR spectra of 15 were achieved by 2D NMR spectra.

**1. Introduction.** – Macrocyclic lactams, derived from spermidine or spermine, are ubiquitous in nature. They have attracted the interest of organic chemists as synthetic targets due to a broad spectrum of biological activity [1]. Dihydroperiphylline (15) is an alkaloid isolated from leaves of the New Caledonian endemic plant Peripterygia *marginata* [2]. The single chiral center in **15** was assumed to have  $(S)$ -configuration. Racemic 15 was synthesized by *Wasserman* and *Matsuyama* [3]. The first enantioselective route, based on a 15-step preparation of  $N-Boc-(S)-\beta$ -phenyl- $\beta$ -alanine from diethyl L-tartrate, was performed by Kaseda et al. [4], confirming the  $(S)$ -configuration. More recently, alternative syntheses of  $(S)$ -15 by cyclization of methyl  $(S)$ -12-amino-4,8-diazadodecanoate in the presence of  $(Me_2N)_3B$  [5] or by ring expansion of  $(S)$ -4phenyl-2-azetidinone [6] were reported. In the present paper, we report a novel, short, and highly efficient approach to 15, based on metal-templated macrocyclization as the key step.

**2. Results and Discussion.**  $-$  Our synthetic strategy is outlined in *Schemes 1* and 2. A highly stereoselective preparation of the chiral  $\beta$ -phenyl- $\beta$ -alanine derivative 3 was performed according to the method of *Davies et al.* [7]. Thus, addition of the *in situ* generated Li derivative of the optically pure amine 1 (prepared from commercially available  $(R)$ - $\alpha$ -methylbenzylamine, 94% ee, *Fluka* [8]) to *tert*-butyl 3-phenylpropenoate 2 (prepared by a modification of the published procedure [9]) gave the corresponding  $\beta$ -phenyl- $\beta$ -alanine derivative 3 in nearly quantitative yield. Both the Nbenzyl- and N- $\alpha$ -methylbenzyl groups were removed by catalytic hydrogenation [7b] in the presence of *Pearlman*'s catalyst, and the deprotected amino ester 4 was isolated in 63% yield after column chromatography. In addition, two side-products were isolated: tert-butyl 3-phenylpropanoate  $(5, 14\%)$  and the aminoester 6 (13%). Formation of the latter indicates the much higher resistance of the  $N-\alpha$ -methylbenzyl residue to reductive removal compared to unsubstituted N-benzyl group, a finding that is

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consistent with the observation of Li et al. [8]. The amino ester 4 was obtained in 92% ee, which was determined by <sup>1</sup>H-NMR *via* derivatization with *Mosher's* (*R*)-acid chloride  $[10]$  (see *Exper. Part*). The amino ester 4 was tosylated to give 7 (84% yield). Subsequent ester hydrolysis was quantitatively accomplished with  $CF_3COOH$  (TFA) at ambient temperature to yield the chiral building block 8 (53% overall yield from 2; Scheme 1).



a) 1. BuLi, THF,  $-80^{\circ}$ ; quant. b) H<sub>2</sub>, 3.8 bar, 10% Pd(OH)<sub>2</sub>/C, EtOH; 63%. c) TsCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>,  $-10^{\circ}$ ; 84%. d) TFA, r.t.; quant. e) Boc<sub>2</sub>O, dioxane, r.t.; 85%. f) TsCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>,  $-10^{\circ}$ ; 87%. g) 1. TFA, CH<sub>2</sub>Cl<sub>2</sub>, r.t.; 2. aq. Na<sub>2</sub>CO<sub>3</sub>; 90%.

The synthesis of 15 also required the preparation of monotosylated butane-1,4 diamine 11. Difficulties arising during a direct monoderivatization of putrescine, however, are well-documented [11]. Putrescine tends to react at both amino groups to give symmetric bifunctional derivatives even in the presence of a large excess of the diamine relative to the functionalizing reagent. However, N-Boc derivatives have been prepared from terminal aliphatic diamines in high yields [12]. Following the procedure in  $[12]$ , we prepared N-Boc-butane-1,4-diamine (9) by reaction of Boc<sub>2</sub>O with an excess of putrescine in dioxane (Scheme 1). The reaction of 9 with TsCl afforded N-Boc-N'-(toluene-4-sulfonyl)butane-1,4-diamine (10). The Boc group was finally cleaved with TFA to give the compound 11.

To perform the coupling between the two key components, 8 and 11, the protected amino acid 8 was first converted to the corresponding acyl chloride by the action of an excess of SOCl<sub>2</sub> in the presence of catalytic amounts of DMF. Subsequent reaction with 11 in the presence of  $Et_3N$  gave the amide 12 in 85% yield (*Scheme 2*).



a) 1. SOCl<sub>2</sub>, DMF (cat.); 2. 11, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>;  $-10^{\circ}$ ; 85%. b) X(CH<sub>2</sub>)<sub>3</sub>X, Cs<sub>2</sub>CO<sub>3</sub>, DMF; X = OTs, 50°, 24 h; 56%; X = OMs, r.t., 72 h; 78%). c) electrolysis, EtOH/DMF; quant. d) PhCH=CHCOCl, DMAP, CH<sub>2</sub>Cl<sub>2</sub>,  $-80^{\circ}$ ; 94%.

Cyclization reactions, mediated by alkali-metal carbonates in polar aprotic solvents like DMF or MeCN are known to be effective for the synthesis of a variety of macrocyclic compounds [13]. This strategy was also successfully applied in our laboratory to obtain some macrocyclic lactams in moderate to high yields [14]. In the present case, slow addition of a solution of equimolar amounts of 1,3-bis(tosyloxy)propane [15] and 12 to a suspension of  $Cs_2CO_3$  in DMF at 50 $\degree$  resulted in the formation of the desired macrocycle 13 in 56% yield. When the more reactive 1,3 bis(methylsulfonyl)propane [16] was used, cyclization proceeded even at ambient temperature, and the yield of 13 increased to 78%. It should be noted that the order of addition is important. For example, slow addition of the alkylating agent to a mixture of  $Cs<sub>2</sub>CO<sub>3</sub>$  and 12 resulted in a lower yield of 13.

Among various methods for deprotecting tosyl-protected amines [17], electrochemical reduction was shown to be the most convenient procedure [18]. In our case, voltage-controlled electrolysis of 13 gave the corresponding macrocyclic lactam 14 in quantitative yield. Finally, 14 was selectively acylated [5] with 1 equiv. of 3-phenylpropenoyl chloride in the presence of 4-(dimethylamino)pyridine (DMAP) at low temperature to afford  $(S)$ -dihydroperiphylline (15) in 94% yield. No acylation at the more hindered N-atom was observed, and the spectral and optical properties of 15 were in agreement with previously published data [4] [5].

In the <sup>1</sup>H-NMR spectra of **15** in CDCl<sub>3</sub> or  $(D_6)$ DMSO, many signals were broad at ambient temperature due to restricted rotations in the disubstitued cinnamoyl amide, and in the 13C-NMR spectra many signals were either broad or doubled. This problem could be overcome by recording the NMR spectra at 373 K in  $(D_6)$ DMSO. Using two dimensional  ${}^{1}H,{}^{1}H$  (COSY) and  ${}^{1}H,{}^{13}C$  (HMBC, HSQC) correlation spectra, we were able to make unambiguous assignments of all  $^1H$  and  $^{13}C$  signals (see *Exper. Part*).

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## Experimental Part

General. All chemicals and solvents were obtained from commercial sources (Fluka, Aldrich, Merck) and used without further purification unless stated otherwise. Technical-grade solvents were distilled prior to use. DMF was stored over flame-dried molecular sieves  $(4 \text{ Å})$ . THF was distilled over Na/benzophenone.  $(R)$ -N-Benzyl-a-methylbenzylamine [8], 1,3-bis(tosyloxy)propane [15] and 1,3-bis(methylsulfonyloxy)propane [16] were prepared according to published procedures. CC:  $SiO_2$ , Merck 60 (40-63 µm). TLC: precoated  $SiO_2$ plates, Merck 60  $F_{254}$ ; detection by UV at 254 nm; Fluram reagent (Fluka) in acetone, fluorescence at 366 nm (for primary amines), Schlittler reagent [19] (for amines and polyamines). M.p.: Mettler FP-5. ORD: Perkin-Elmer 241 polarimeter. IR (cm<sup>-1</sup>): *Perkin-Elmer Spectrum One*; KBr pellets or neat substance. <sup>1</sup>H-NMR: Bruker ARX-300 (300 MHz) or AMX-600 (600 MHz); chemical shifts  $\delta$  in ppm relative to Me<sub>4</sub>Si as internal standard; J values in Hz. <sup>13</sup>C-NMR: *Bruker ARX-300* (75 MHz) or  $AMX-600$  (150 MHz); chemical shifts  $\delta$  in ppm relative to Me<sub>4</sub>Si, signals multiplicity determined from DEPT experiments. CI-MS (NH<sub>3</sub> as reactant gas): Finnigan MAT 90. ESI-MS (NaI/MeOH/CH<sub>2</sub>Cl<sub>2</sub>): Finnigan TSQ 700; m/z and relative intensities (% of base peak) are given.

tert-Butyl (E)-3-Phenylprop-2-enoate (2). 3-Phenylprop-2-enoyl chloride (49.8g, 0.30 mol) was added in portions to a vigorously stirred mixture of  $t$ -BuOH (32.8 ml, 0.35 mol) and Et<sub>3</sub>N (48.4 ml, 0.45 mol). The mixture was stirred at 90 $^{\circ}$  for 12 h (a white solid precipitated), then cooled to r.t., Et<sub>2</sub>O (100 ml) and H<sub>2</sub>O (50 ml) were added, and stirring was continued until all the solid was dissolved. The org. layer was separated, washed with 1N aq. HCl ( $3 \times 50$  ml), sat. aq. NaHCO<sub>3</sub> (50 ml), H<sub>2</sub>O (50 ml), dried (MgSO<sub>4</sub>) and concentrated. Distillation of the residue in vacuo afforded 44.8 g (73%) of 2. Colorless liquid. B.p. 72 – 73°/0.1 Torr ( $[20]$ : 80 –  $81\degree/0.3$  Torr). For IR, MS,  $^1$ H- and  $^{13}$ C-NMR spectra, see [20].

tert-Butyl (3S,aR)-3-[N-Benzyl-N-(1-phenylethyl)amino]-3-phenylpropanoate (3). A stirred soln. of (R)- $N$ -benzyl- $\alpha$ -methylbenzylamine (15.8 g, 75 mmol) in dry THF (150 ml) was cooled to  $-80^{\circ}$  and treated dropwise with BuLi (1.6M soln. in hexane, 44 ml, ca. 70 mmol). The resulting red soln. was stirred for 1 h. Then, a soln. of 2 (10.2 g, 50 mmol) in dry THF (40 ml) was added dropwise. Stirring was continued for 1 h. Sat. aq. NH<sub>4</sub>Cl (60 ml) was added, and the mixture was allowed to reach r.t. THF was removed in vacuo, and the residue was partitioned between H<sub>2</sub>O (50 ml) and CH<sub>2</sub>Cl<sub>2</sub> (150 ml). The org. phase was washed with 10% citric acid  $(2 \times 100 \text{ ml})$ , H<sub>2</sub>O (50 ml), dried (MgSO<sub>4</sub>), and evaporated to give nearly pure 3 (20.8 g, 100%), which was used without further purification. Yellow oil.  $R_f$  0.82 (CH<sub>2</sub>Cl<sub>2</sub>). IR (neat): 3085m, 3002m, 3062m, 3028s, 2976s, 2932m, 1948w, 1877w, 1807w, 1727s, 1601m, 1493s, 1478m, 1453s, 1367s, 1301m, 1237m, 1152s, 1083m, 1057w, 1028m, 962m, 912m, 749s, 699s, 647w, 549w. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 1.20 (s, t-Bu); 1.24 (d, J = 6.8, Me); 2.48 (dd, J = 14.6, 9.4, 1 H, CH<sub>2</sub>CO); 2.55 (dd, J = 14.6, 5.8, 1 H, CH<sub>2</sub>CO); 3.67 (s, PhCH<sub>2</sub>); 3.98 (q, J = 6.8, MeCH); 4.40 (dd, J = 9.4, 5.8, CH<sub>2</sub>CH); 7.15 - 7.32 (m, 15 arom. H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 16.34 (q, Me); 27.77 (q, Me<sub>3</sub>C); 38.50 (t, CH<sub>2</sub>CO); 50.86 (t, PhCH<sub>2</sub>); 57.10, 59.68 (2d, 2 CH); 80.05 (s, Me<sub>3</sub>C); 126.45, 126.76, 127.03 (3d, 3 arom. CH); 127.80 -128.25 (several d, 12 arom. CH); 141.66, 141.80, 144.15 (3s, 3 arom. C); 171.03 (s, C=O). CI-MS: 416 (100,  $[M +]$  $1$  $\uparrow$ , 300  $(12)$ .

tert-Butyl (S)-3-Amino-3-phenylpropanoate (4). A soln. of 3 (20.8 g, 50 mmol) in dry EtOH (200 ml) was placed in a pressure-resistant bottle. The bottle was purged with  $N_2$ , and Pearlman's catalyst (10% Pd(OH)<sub>2</sub> on charcoal, 4.0 g) was added. The pressure head was adjusted, the bottle was placed in a Parr apparatus, evacuated, purged with  $N_2$ , and, then, 3 times with H<sub>2</sub>. The pressure was set to 3.8 bar, and the bottle was vigorously shaken for 72 h. Then, the pressure was released, the mixture was filtered through Celite, and EtOH was evaporated. CC  $(CH_2Cl_2, then CH_2Cl_2/MeOH 20:1)$  afforded 4 (6.96 g, 63%) as well as 5 (1.48 g, 14%) and 6 (2.11 g, 13%).

*Data of* 4: colorless oil.  $R_f$  0.60 (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 10:1). [ $\alpha$ ]<sub>D</sub> = -19.3 ( $c$  = 1.06, CHCl<sub>3</sub>). IR (neat): 3382*m*, 3314m, 3063m, 3029m, 2978s, 2931m, 1953w, 1725s, 1604m, 1494m, 1479w, 1454m, 1392m, 1368s, 1319m, 1256m, 1209m, 1150s, 1028w, 1015w, 955m, 909w, 846m, 755m, 700s, 593w. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 1.41 (s, t-Bu); 1.77 (br. s, NH<sub>2</sub>); 2.58 (d, J = 6.9, CH<sub>2</sub>CO); 4.37 (t, J = 6.9, CHN); 7.24 - 7.37 (m, 5 arom. H). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 27.95  $(q, C(CH_3)_{3})$ ; 45.23 (t, CH<sub>2</sub>); 52.69 (d, CHN); 80.60 (s, C(CH<sub>3</sub>)<sub>3</sub>); 126.20 (d, 2 arom. CH); 127.16 (d, 1 arom. CH); 128.41 (d, 2 arom. CH); 144.64 (s, 1 arom. C); 171.20 (s, C=O). CI-MS: 443 (24,  $[2M+1]^+$ ), 222 (100,  $[M+1]^+$ ). <sup>1</sup>H-NMR of *Mosher*'s amide of **4** (selected signals): (*S*,*S*)-isomer: 3.32 (*q*, <sup>5</sup>*J*(H,F) = 1.4, OCH<sub>3</sub>);  $(S,R)$ -isomer: 3.40  $(q, {}^{5}J(H,F) = 1.4, OCH_3)$ ; integral ratio  $(S,S)/(S,R) = 23.6$ : 1, corresponding to ee 92%.

*Data of* 5: colorless liquid. <sup>1</sup>H-NMR: 1.41 (s, t-Bu); 2.54 (t, J = 7.8, CH<sub>2</sub>CO); 2.91 (t, J = 7.8, PhCH<sub>2</sub>); 7.18 – 7.27 (*m*, 5 arom. H). CI-MS: 224 (72,  $[M + NH_4]^+$ ), 168 (100,  $[(M - C_4H_8) + NH_4]^+$ ).

Data of 6: colorless oil. R<sub>f</sub> 0.62 (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 20:1). IR (neat): 3328w, 3062w, 3028m, 2976m, 2930m, 1950w, 1726s, 1602w, 1493m, 1453m, 1367s, 1296m, 1258m, 1151s, 1079w, 1028w, 957m, 911w, 8 45m, 761m, 699s, 665w. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 1.34 (d, J = 6.5, Me); 1.36 (s, t-Bu); 1.85 (br. s, NH); 2.59 (dd, J = 10.7, 7.1, CH<sub>2</sub>); 3.66  $(q, J = 6.5, \text{MeCH})$ ; 4.15  $(dd, J = 7.7, 6.3, \text{CHN})$ ; 7.18 – 7.32  $(m, 10 \text{ arcm. H})$ . <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 22.19  $(q, \text{Me})$ ; 27.90  $(q, Me_3C)$ ; 43.84  $(t, CH_2)$ ; 54.48, 57.03 (2t, 2 CHN); 80.35  $(s, Me_3C)$ ; 126.46 - 127.02 (several d, 6 arom. CH); 128.20, 128.27 (2d, 2 × 2 arom. CH); 142.77, 145.97 (2s, 2 arom. C); 170.86 (s, C=O). CI-MS: 326  $([M+1]^+)$ .

tert-Butyl (S)-3-{[(4-Methylphenyl)sulfonyl]amino}-3-phenylpropanoate (7). A stirred soln. of 4 (3.09 g, 14 mmol) and Et<sub>3</sub>N (2.8 ml, 20 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (50 ml) was cooled to  $-10^{\circ}$  and treated dropwise with TsCl  $(2.67 \text{ g}, 14 \text{ mmol})$  in CH<sub>2</sub>Cl<sub>2</sub> (20 ml), stirred for 30 min at  $-10^{\circ}$  and another 30 min at r.t. The resulting mixture was washed with 10% citric acid (20 ml), H<sub>2</sub>O (2  $\times$  20 ml), dried (MgSO<sub>4</sub>), and concentrated in vacuo. CC  $(CH_2Cl_2/ACOEt 20:1)$  afforded **7** (4.43 g, 84%). White solid.  $R_f$  0.52  $(CH_2Cl_2/ACOEt 20:1)$ . M.p. 71 – 72°.  $[\alpha_p =$ -35.6 (c 1.0, CHCl3). IR (KBr): 3250s, 3033m, 3062m, 2988s, 2753w, 2642w, 2293w, 1900w, 1717s, 1686s, 1600m, 1496m, 1456s, 1427s, 1326s, 1303s, 1252m, 1207m, 1159s, 1096s, 1060s, 968s, 8 58m, 8 09m, 771m, 703s, 665s, 603*m*, 555s, 534s, 467*m*. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 1.30 (s, *t*-Bu); 2.34 (s, Me); 2.63 (dd, J = 15.4, 6.5, 1 H, CH<sub>2</sub>CO); 2.73  $(dd, J=15.4, 6.5, 1$  H, CH<sub>2</sub>CO); 4.70  $(dt, J=7.8, 6.5,$  CHN); 5.97  $(d, J=7.8,$  NH); 7.08 - 7.18  $(m, 5$  H of Ph, 2 H of Ts); 7.57 (d,  $J = 8.3$ , 2 H of Ts). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 21.30 (q, Me); 27.75 (q, Me<sub>3</sub>C); 42.36 (t, CH<sub>2</sub>); 54.61  $(d, \text{CHN})$ ; 81.47 (s, Me<sub>3</sub>C); 126.49, 126.97 (2d, 2 × 2 arom. CH); 127.43 (d, 1 arom. CH); 129.24, 130.13 (2d, 2 × 2 arom. CH); 137.55, 139.33, 142.92 (3s, 3 arom. C); 169.82 (s, C=O). CI-MS: 393 (18,  $[M + NH_4]^+$ ), 337 (100,  $[(M - C_4H_8) + NH_4]^+$ , 189 (6).

(S)-3-{[(4-Methylphenyl)sulfonyl]amino}-3-phenylpropanoic Acid (8). A mixture of 7 (2.14 g, 5.71 mmol) and TFA (5 ml) was stirred at r.t. for 3 h. Evaporation of volatile materials and drying of the residue in vacuo afforded **8** (1.82 g, 100%). White solid.  $R_f$  0.31 (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 10:1). M.p. 148–150°. [ $a$ ]<sub>1</sub> $= -33.1$  ( $c = 1.0$ , MeOH). IR (KBr): 3270s, 3063m, 3028m, 2980m, 2917m, 2679m, 2577m, 1960w, 1907w, 1710s, 1598m, 1495m, 1460m, 1447m, 1414m, 1379w, 1352m, 1326s, 1306s, 1211m, 1162s, 1086m, 1064m, 1028w, 1019w, 955m, 8 43m, 810m, 763m, 703s, 670s, 616m, 573m, 551m, 527m, 477m. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 2.37 (s, Me); 2.81 (dd, J = 16.4, 6.2, 1 H, CH<sub>2</sub>); 2.92 (dd, J = 16.4, 6.2, 1 H, CH<sub>2</sub>); 4.73 (dt, J = 7.6, 6.2, CHN); 5.67 (d, J = 7.6, NH); 7.08 – 7.21 (m, 5 H of Ph, 2 H of Ts); 7.59 (d,  $J = 8.3$ , 2 H of Ts); 11.5 – 12.5 (very br. s, COOH). <sup>13</sup>C-NMR ((D<sub>6</sub>)DMSO): 20.74  $(q, \text{Me})$ ; 42.00  $(t, \text{CH}_2)$ ; 54.38  $(d, \text{CHN})$ ; 126.19, 126.51 (2d, 2 × 2 arom. CH); 126.86  $(d, 1 \text{ arom. CH})$ ; 127.86, 128.96 (2d, 2 × 2 arom. CH); 138.38, 140.57, 141.91 (3s, 3 arom. C); 170.86 (s, C=O). ESI-MS: 342 ( $[M + Na]$ <sup>+</sup>).

tert-Butyl (4-[[(4-Methylphenyl)sulfonyl]amino]butyl)carbamate (10). A stirred soln. of 9 (4.00 g, 21.3 mmol) and Et<sub>3</sub>N (5.5 ml, 40 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (100 ml) was cooled to  $-10^{\circ}$ , and treated dropwise with a soln. of TsCl (4.19 g, 22 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 ml). The mixture was stirred for 30 min at 0° and another 30 min at r.t. The resulting mixture was washed with 10% citric acid (25 ml), H<sub>2</sub>O ( $2 \times 25$  ml), dried (MgSO<sub>4</sub>), and evaporated. Extraction of the residue with hexane  $(3 \times 50 \text{ ml})$  gave 7.25 g of crude product, which solidified upon standing. Crystallization from hexane/AcOEt gave pure 10 (6.30 g, 87%). White solid.  $R_f$  0.19  $(CH_2Cl_2/ACOEt \ 10:1)$ . M.p.  $85-87^{\circ}$  ([21]:  $87.5-88.5^{\circ}$ ). For IR, <sup>1</sup>H- and <sup>13</sup>C-NMR spectra, see [21]. ESI-MS: 365 (84,  $[M + Na]$ <sup>+</sup>), 309 (39,  $[(M - C_4H_8) + Na]$ <sup>+</sup>), 265 (10,  $[(M - C_4H_8 - CO_2) + Na]$ <sup>+</sup>), 243 (100,  $[(M - C_4H_8 - CO_2) + 1]^+$ .

 $N-(4-Aminobutyl)-4-methylbenzenesulfonamide (11)$ . A stirred soln. of 10 (3.42 g, 10 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (25 ml) was treated with TFA (3.8ml, 50 mmol) in one portion and stirred for 1 h at r.t. The acid was evaporated, the residue was treated with 10% aq. Na<sub>2</sub>CO<sub>3</sub> (50 ml) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (5  $\times$  75 ml). The combined extracts were dried  $(MgSO<sub>4</sub>)$  and evaporated to afford 11 (2.15 g, 89%), which solidified upon standing. Slightly yellow solid.  $R_f$  0.41 (CH<sub>2</sub>Cl<sub>2</sub>/MeOH/25% aq. NH<sub>3</sub> 70 : 30 : 3). M.p. 64 – 66 $^{\circ}$  ([22]: 65 – 68 $^{\circ}$ ). IR

(neat): 3583m, 3362m, 3276s, 3064s, 2936s, 1867s, 1924w, 1688m, 1598m, 1495m, 1452m, 1323s, 1201m, 1157s,  $1094$ s,  $1019$ w,  $916$ w,  $816$ m,  $737$ w,  $707$ m,  $660$ s,  $618$ w,  $572$ m,  $551$ s.  $\rm{^1H\text{-}NMR}$  (CDCl<sub>3</sub>):  $1.38-1.58$   $(m,2 \text{ }CH_2CH_2N)$ ; 2.42 (s, Me); 2.66 (t, J = 6.3, CH<sub>2</sub>N); 2.92 (t, J = 6.5, CH<sub>2</sub>N); 3.8 (br. s, 3 H, NH<sub>2</sub> + NH); 7.29 (d, J = 8.1, 2 arom. H); 7.74  $(d, J = 8.1, 2 \text{ arom. H})$ . <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 21.33  $(q, Me)$ ; 27.05, 29.53  $(2t, 2 \text{ CH}_2\text{CH}_2\text{N})$ ; 40.90, 42.80 (2t, 2 CH<sub>2</sub>N); 126.89, 129.49 (2d, 2 × 2 arom. CH); 137.17, 142.95 (2s, 2 arom. C). ESI-MS: 485 (38, [2M +  $1$ <sup>+</sup>), 265 (21,  $[M + Na]$ <sup>+</sup>), 243 (100,  $[M + 1]$ <sup>+</sup>).

(S)-3-[(4-Methylphenyl)sulfonylamino]-N-{4-[(4-methylphenyl)sulfonyl]aminobutyl}-3-phenylpropanamide (12). To 8 (1.276 g, 4 mmol) was added  $SOCl<sub>2</sub>$  (4 ml) followed by DMF (ca. 0.05 ml). The resulting mixture was stirred at r.t. for  $2 h$ . The excess of SOCl, was evaporated to give the corresponding acyl chloride which was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (20 ml) and added dropwise to stirred soln. of 11 (1.09 g, 4.5 mmol) and Et<sub>3</sub>N (0.83 ml, 6 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (30 ml) at  $-10^{\circ}$ . Stirring was continued for 30 min at  $-10^{\circ}$  and for another 30 min at r.t. The resulting soln. was washed with 10% citric acid (20 ml), H<sub>2</sub>O ( $2 \times 20$  ml), dried (MgSO<sub>4</sub>) and concentrated in vacuo. Purification of the residue by CC (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 20:1) afforded 12 (1.85 g, 85%). White solid.  $R_f$  0.57 (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 10:1). M.p. 138–139°. [ $\alpha$ ]<sub>0</sub> = -26.5 ( $c$  = 1.00, MeOH). IR (KBr): 3744w, 3287m, 2928m, 2866w, 1914w, 1805w, 1647s, 1598m, 1542s, 1495m, 1455m, 1423m, 1322s, 1211w, 1155s, 1092s, 1067m, 967w, 842w, 813m, 744w, 703m, 671s, 604w, 573m, 551s. 1H-NMR (CDCl<sub>3</sub>): 1.44 (m, 2 CH<sub>2</sub>CH<sub>2</sub>N); 2.30, 2.41 (2s,  $2 \text{ Me}$ ); 2.50 (dd, J = 14.4, 5.2, 1 H, CH<sub>2</sub>CO); 2.58 (dd, J = 14.4, 8.0, 1 H, CH<sub>2</sub>CO); 2.85 (m, CH<sub>2</sub>NHTs); 3.03  $(m, 1 H, CH, NHCO)$ ; 3.21  $(m, 1 H, CH, NHCO)$ ; 4.74  $(ddd, J = 8.1, 5.2, 8.0, CHN)$ ; 5.42  $(t, J = 6.1, CH, NHTs)$ ; 6.30 (t,  $J = 5.7$ , CH<sub>2</sub>NHCO); 6.72 (d,  $J = 8.1$ , CHNHTs); 7.01 – 7.06 (m, 5 H of Ph, 2 H of Ts); 7.29 (d,  $J = 8.0$ , 2 H of Ts); 7.48 (d,  $J = 8.3$ , 2 H of Ts); 7.73 (d,  $J = 8.3$ , 2 H of Ts). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 21.26, 21.38 (2q, 2 Me); 26.14, 26.40 (2t, 2 CH2CH2N); 38.66 (t, CH2CO); 42.77, 43.38(2t, 2 CH2N); 55.43 (d, CHN); 126.39, 126.89, 126.96 (3d,  $3 \times 2$  arom. CH); 127.15 (d, 1 arom. CH); 128.18, 129.10, 129.64 (3d,  $3 \times 2$  arom. CH); 136.54, 137.45, 139.75, 142.74, 143.33 (5s, 5 arom. C); 170.09 (s, C=O). CI-MS: 561 (45,  $[M + NH_4]^+$ ), 544 (81,  $[M + 1]^+$ ), 390 (38,  $[(M - TsNH<sub>2</sub>) + NH<sub>4</sub>]<sup>+</sup>$ , 373 (100,  $[(M - TsNH<sub>2</sub>) + 1]<sup>+</sup>$ ), 189 (80).

(S)-5,9-Bis[(4-methylphenyl)sulfonyl]-4-phenyl-1,5,9-triazacyclotridecan-2-one (13). Method A. A stirred suspension of Cs<sub>2</sub>CO<sub>3</sub> (0.717 g, 2.2 mmol) in dry DMF (30 ml) was warmed to 50 $^{\circ}$  and treated dropwise with a soln. of 12 (0.543 g, 1.0 mmol) and 1,3-bis(tosyloxy)propane (0.384 g, 1.0 mmol) in dry DMF (40 ml) during 2 h. After the addition was completed, stirring was continued for 24 h, then the reaction mixture was cooled to r.t. The solvent was evaporated in vacuo and the residue was partitioned between  $CH_2Cl_2(50 \text{ ml})$  and  $H_2O(20 \text{ ml})$ . The org. layer was washed with H<sub>2</sub>O ( $2 \times 20$  ml), dried (MgSO<sub>4</sub>), and concentrated in vacuo. CC of the residue  $(CH_2Cl_2/ACOE14:1)$  afforded 13 (0.328 g, 56%). White foam.  $R_f$  0.53 (CH<sub>2</sub>Cl<sub>2</sub>/AcOEt 1:1).  $[a]_D = +1.3$  ( $c =$ 1.27, CHCl3). IR (KBr): 3380m, 3062w, 3030w, 2927m, 2866m, 1918w, 1751m, 1667s, 1598m, 1535m, 1495m, 1452m, 1335s, 1215m, 1155s, 1121m, 1089m, 1035w, 996w, 952w, 919w, 815m, 763m, 701m, 726m, 687m, 655m, 607w, 568m, 549s. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 1.42, 1.56 (2m, 2 H, CH<sub>2</sub>CH<sub>2</sub>N); 1.76 - 2.03 (m, 3 H, CH<sub>2</sub>CH<sub>2</sub>N); 2.11  $(m, 1 \text{ H}, \text{CH}_2\text{CH}_2\text{N})$ ; 2.26, 2.41 (2s, 2 Me); 2.69 (dd, J = 14.6, 2.0, 1 H, CH<sub>2</sub>CO); 2.92 (m, 1 H, CH<sub>2</sub>NHCO); 3.11 - 3.38 (m, 6 H, 3 CH<sub>2</sub>NHTs); 3.67 (dd, J = 14.6, 11.6, 1 H, CH<sub>2</sub>CO); 3.90 (m, 1 H, CH<sub>2</sub>NHCO); 5.07  $(dd, J=11.6, 2.0, CHN); 6.26 (dd, J=8.6, 3.4, CH<sub>2</sub>NHCO); 6.89 (d, J=8.3, 2 H of Ts); 6.98-7.21 (m, 5 H of Ph,$ 2 H of Ts); 7.28 (d, J = 8.4, 2 H of Ts); 7.68 (d, J = 8.3, 2 H of Ts). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 21.16, 21.34 (2q, 2 Me); 22.42, 24.73, 28.52 (3t, 3 CH<sub>2</sub>CH<sub>2</sub>N); 38.48 (t, CH<sub>2</sub>CO); 41.86, 44.28, 46.75, 48.03 (4t, 4 CH<sub>2</sub>N); 62.21 (d, CHN); 127.03 ± 128.29 (several d, 9 arom. CH); 128.78, 129.53 (2d, 2 2 arom. CH); 136.65, 136.92, 137.73, 142.47, 142.88 (5s, 5 arom. C); 170.64 (s, C=O). ESI-MS: 1189 (17,  $[2M + Na]^+$ ), 606 (100,  $[M + Na]^+$ ), 584 (13,  $[M + Na]^+$ )  $1]^+$ 

Method B. A stirred suspension of  $Cs_2CO_3$  (0.717 g, 2.2 mmol) in dry DMF (30 ml) was treated dropwise at r.t. with a soln. of 12 (0.543 g, 1.0 mmol) and 1,3-bis(methylsulfonyloxy)propane (0.384 g, 1.0 mmol) in dry DMF (40 ml) during 2 h. After the addition was complete, stirring was continued for 72 h. The same workup as above afforded 13 (0.457 g, 78%).

(S)-4-Phenyl-1,5,9-triazacyclotridecan-2-one (14). A soln. of 13 (0.419 g, 0.72 mmol) in a minimal amount of DMF was added to a 0.1M soln. of Me<sub>4</sub>NCl in EtOH (100 ml). The resulting soln. was subjected to electrolysis under Ar at -2.25 V in a three-electrode cell with a Hg cathode, graphite-rod anode, and standard calomel electrode as reference electrode (see [18] for detailed procedure). After the electrolysis was complete, the solvents were removed in vacuo, the remainder was dissolved in 20% aq.  $K_2CO_3$  (10 ml) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (4  $\times$  50 ml). The combined extracts were dried and evaporated to afford 14 (0.198 g, 100%). Colorless oil.  $R_f$  0.20 (CH<sub>2</sub>Cl<sub>2</sub>/MeOH/25% aq. NH<sub>3</sub> 70:30:3). [ $\alpha$ ]<sub>D</sub> = -5.2 ( $c$  = 0.89, CHCl<sub>3</sub>). IR (KBr): 3269s, 3025m, 3061m, 2927s, 2855s, 2803m, 1952w, 1881w, 1812w, 1657 (sh), 1636s, 1554s, 1491m, 1453m, 1436m, 1356m, 1308m, 1272w, 1233m, 1185m, 1130m, 1069m, 1028m, 960m, 925w, 8 44w, 799w, 763m, 703s, 646w, 619w, 595m, 569w, 543m, 521m. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 1.69 – 1.88 (m, 6 H, 3 CH<sub>2</sub>CH<sub>2</sub>N); 2.38 – 2.54 (m, CH<sub>2</sub>); 2.60 – 2.84 (m, 4 H,

 $CH_2 + 2 NH$ ; 3.00  $-3.10$  (m, CH<sub>2</sub>); 3.48  $-3.69$  (m, 2 CH<sub>2</sub>N); 4.01 (dd, J = 11.5, 2.8, CHN); 7.21  $-7.35$  (m, 5 H, Ph); 8.52 (br. t, CONH). 13C-NMR: 26.33, 26.61, 27.41 (3t, 3 CH2CH2N); 39.53 (t, CH2CO); 44.33, 45.14, 48.19, 49.00 (4t, 4 CH<sub>2</sub>N); 59.99 (d, CHN); 126.23 (d, 2 arom. CH); 127.18 (d, 1 arom. CH); 128.56 (d, 2 arom. CH); 142.71 (s, 1 arom. C); 171.84 (s, C=O). CI-MS: 276 (100,  $[M+1]^+$ ), 91 (22,  $[C_7H_7]^+$ ).

 $(S)$ -4-Phenyl-9-(3-phenylprop-2-enoyl)-1,5,9-triazacyclotridecan-2-one  $(=(S)$ -Dihydroperiphylline; 15). A stirred soln. of 14 (0.169 g, 0.61 mmol) and DMAP (0.223 g, 1.83 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (15 ml) was cooled to  $-80^{\circ}$ and treated dropwise with a soln. of cinnamoyl chloride (101.5 mg, 0.60 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (5 ml). The mixture was stirred for 1 h at  $-80^{\circ}$ , allowed to reach r.t., washed with H<sub>2</sub>O (2  $\times$  10 ml) and concentrated. CC of the residue (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 10 : 1) afforded **15** (230 mg, 94%). White solid.  $R_f$  0.40 (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 10 : 1). M.p. 81 –  $82^{\circ}$  ([4]: 82.5–83.5 $^{\circ}$ ). [ $a]_p = +3.2$  ( $c = 0.5$ , CHCl<sub>3</sub>). IR (KBr): 3744w, 3400m, 3296m, 3060m, 3026m, 2926m, 2856m, 1952w, 1885w, 1825w, 1646s, 1597s, 1552m, 1495m, 1452s, 1427s, 1375m, 1327m, 1304m, 1254w, 1192m, 1174m, 1114m, 1073m, 1028w, 976m, 911w, 855m, 763s, 702s, 588w, 555m, 534m. <sup>1</sup>H-NMR ((D<sub>6</sub>)DMSO, 373 K):  $1.48-1.62/1.70-1.88$  (2m, CH<sub>2</sub>(7), CH<sub>2</sub>(11), CH<sub>2</sub>(12)); 2.24 (ddd, J = 12.4, 8.8, 3.7, 1, H - C(6)); 2.28 (dd, J = 13.3, 4.3, 1 H – C(3)); 2.37 (dd, J = 13.3, 10.4, 1 H – C(3)); 2.51 (m, H – N(5)); 2.55 (ddd, J = 12.4, 7.0, 3.8,  $1 H-C(6)$ ; 3.22, 3.35 (2m, 2 H, CH<sub>2</sub>(13)); 3.47 – 3.66 (m, CH<sub>2</sub>(8), CH<sub>2</sub>(10)); 3.98 (dd, J = 10.4, 4.3, CHN); 7.00  $(d, J = 15.4, CH = CHCO)$ ; 7.21  $(m, 1 \text{ arom. H})$ ; 7.30 – 7.43  $(m, 7 \text{ arom. H})$ ; 7.45  $(d, J = 15.4, CH = CHCO)$ ; 7.61  $(m, 2)$  H of CH=CHPh); 7.72 (br. t, CONH). <sup>13</sup>C-NMR  $((D<sub>6</sub>)$ DMSO, 373 K): 24.64 (2 overlapping t, C(11),  $C(12)$ ); 29.00 (t,  $C(7)$ ); 37.36 (t,  $C(13)$ ); 42.25 (t,  $C(6)$ ); 43.22 (t,  $C(10)$ ); 45.33 (t,  $C(3)$ ); 47.50 (t,  $C(8)$ ); 59.24  $(d, C(4))$ ; 119.07 (d, CH=CHCO); 125.80 (d, 2 arom. CH); 126.06 (d, 1 arom. CH); 127.11, 127.70, 128.10 (3d,  $3 \times 2$  arom. CH); 128.55 (d, 1 arom. CH); 135.02 (s, 1 arom. CH); 139.85 (d, CH=CHCO); 143.85 (s, 1 arom. C); 164.79 (s, CH=CHCO); 170.33 (CONH). CI-MS: 406 (100, [M + 1], 276 (8).

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